

Information sheet for the course Inorganic Chemistry of Materials

University: <i>Alexander Dubček University of Trenčín</i>						
Faculty: <i>Faculty of Industrial Technologies in Púchov</i>						
Course unit code: <i>MI-P-2</i>				Course unit title: <i>Inorganic Chemistry of Materials</i>		
Type of course unit: <i>compulsory</i>						
Planned types, learning activities and teaching methods: <i>Lecture: 2 hours weekly/26 hours per semester of study</i> <i>Seminar: 2 hours weekly/26 hours per semester of study</i> <i>Laboratory tutorial: 2 hours weekly/26 hours per semester of study</i>						
Number of credits: 6						
Recommended semester: <i>1st semester in the 1st year full-time</i> <i>1st semester in the 1st year part-time</i>						
Degree of study: <i>the 1st degree of study (Bachelor's degree)</i>						
Course prerequisites: <i>none</i>						
Assessment methods: <i>Evaluation of course includes partial evaluation; basic characteristic of static of chemical substances, chemical reaction (thermodynamic and kinetic aspects), kind of chemical bond, physical properties of inorganic substances</i>						
Learning outcomes of the course unit: <i>Student profits the survey on the inorganic chemistry and materials.</i>						
Course contents: <ol style="list-style-type: none"> 1. States of chemical substances – basic characteristic 2. Chemical thermodynamic: Enthalpy, free energy, Entropy, Gibbs energy 3. Chemical equilibrium, equilibrium constant 4. Kinetics: rate laws 5. The effect of concentration, temperature and catalyst on reaction rate 6. Acids and basis (Arrhenius, Bronsted and Lewis theory) 7. Protolytic reactions: neutralization and hydrolysis 8. Precipitation reactions: product of solubility 9. Redox reactions: reducing and oxidizing agents, redox potential 10. Reactions of complex formation: complex, chromophore 11. Wave mechanics: wave function, atomic orbitals, electron configurations (the aufbau principle, Hund's rule, the Pauli principle) 12. Physical essence of chemical bond 13. Kinds of chemical bonds 14. Electric, magnetic, optical and thermal properties of inorganic substances 						
Recommended of required reading: <ol style="list-style-type: none"> 1. Jóna E., Ondrušová D., Pajtášová M.: <i>Priemysel'ná anorganická chémia I: Všeobecná časť</i>, 2007 2. Garaj J.: <i>Chémia učebné texty pre mechanické odbory</i>, Trenčín, 						
Language: <i>Slovak</i>						
Remarks: <i>none</i>						
Evaluation history:						
A	B	C	D	E	FX	

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Lecturers: <i>prof. Ing. Eugen Jóna, DrSc., prof. Ing. Darina Ondrušová, PhD.</i>					
Last modification: <i>31.03.2014</i>					
Supervisor: <i>prof. Ing. Darina Ondrušová, PhD.</i>					